## Extending stability of cathode electrode materials in high-voltage

## region for Li/Na-ion batteries

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To further develop high-energy density Li/Na ion batteries, a feasible strategy is to develop cathode materials is able to work at high-voltage. For example, the upper limit voltage of commercial LiCoO<sub>2</sub> has been extended to 4.4 or 4.45(vs. Li). In order to achieve this target, highly stable electrode/electrolyte materials are necessary. In this talk, we will present some results about understanding how to keep high stability of electrode/electrolyte at high voltage with alivolent doping and electrolyte additives.

In the first example<sup>[1]</sup>, suberonitrile (SUN) and LiBOB were investigated as binary additives for Li-ion batteries that used LiCoO<sub>2</sub> as a cathode, cycled at high cutoff potentials and utilized LiPF<sub>6</sub>-based electrolyte. In the electrolyte with binary additives, the electrochemical performance of LiCoO<sub>2</sub> was enhanced significantly, and the initial coulombic efficiency increased to 94% compared with 90% in reference electrolyte. The cell also exhibited capacity retention of 62% after 500 cycles, a strong contrast with the 25% measured in reference electrolyte. EIS, SEM, TEM and XPS analyses indicated that the combination of the two additives had a unique influence on the structure and composition of the cathode-electrolyte interface on LiCoO<sub>2</sub>. Ex situ XRD confirmed that structural change of the LiCoO<sub>2</sub> material was not the main reason causing poor cycling performance when charging to the 4.5 V cutoff potential. In addition, some new progress<sup>[2]</sup> in the controlling cycling stability of P2-type Na<sub>0.66</sub>Ni<sub>0.33</sub>Mn<sub>0.67</sub>O<sub>2</sub> with Zn<sup>2+</sup>-doping and its working mechanism will be also presented.



Figure 1.Comparison of (a) the initial voltage profiles at 14 mAg<sup>-1</sup>, (b) the cycling performance and coulombic efficiencies of LiCoO<sub>2</sub>/Li half cells at140 mAg<sup>-1</sup>.

Reference:

[1] Y. Ji, S. Li, G. Zhong, Z. Zhang, Y. Li, M. J. McDonald, Y. Yang *Journal of The Electrochemical Society*.
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[2] a) X.H.Wu, J.H.Guo, et al, *J Power Sources*, 281(2015), 18-23; b) X.H.Wu, G.M.Zhong et al; to be submitted